

## Principles of the Geological Storage of Carbon Dioxide

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## **Key Messages**

- Evidence from petroleum studies show that oil, gas and CO<sub>2</sub> can be stored in the deep subsurface for geological time
- 2. CO<sub>2</sub> can be stored in depleted oil and gas fields and deep saline formations
- 3. CO<sub>2</sub> injected as a fluid into reservoirs (sandstones) is trapped by seals (mudstones)
- 4. The technology for the geological storage of CO<sub>2</sub> is mature



## What is Geological Storage?

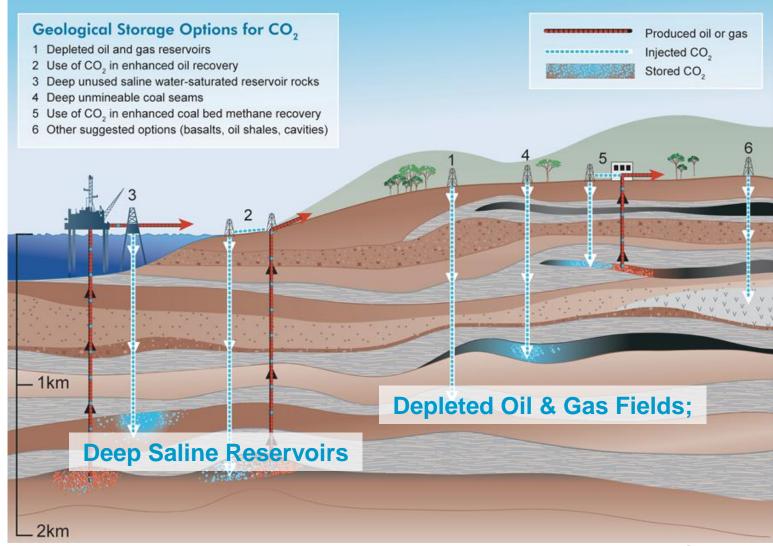
- Carbon dioxide captured from industrial processes is transported to a storage site
- Injection via a well bore into a deep geological formation as a dense (supercritical) fluid
- Monitoring the migration of the fluid under buoyancy away from the injection point
- Eventual the CO2 is permanently trapping – structural (buoyancy), dissolution, residual and geochemical



#### **Options for Geological Storage**

International Energy Agency

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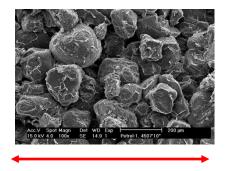


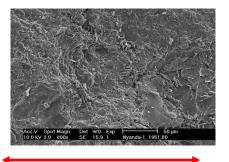
Source IP O CO IEA 2012



## **Reservoirs and Seals**

- Reservoir rocks have spaces (pores) between the grains which can hold fluids and connections between the pores which can allow the fluids to flow through them (permeability). Sandstones and limestones.
- Sealing rocks are very fine grained with no practical permeability. Mudstones or shales.







~1/4 millimetre



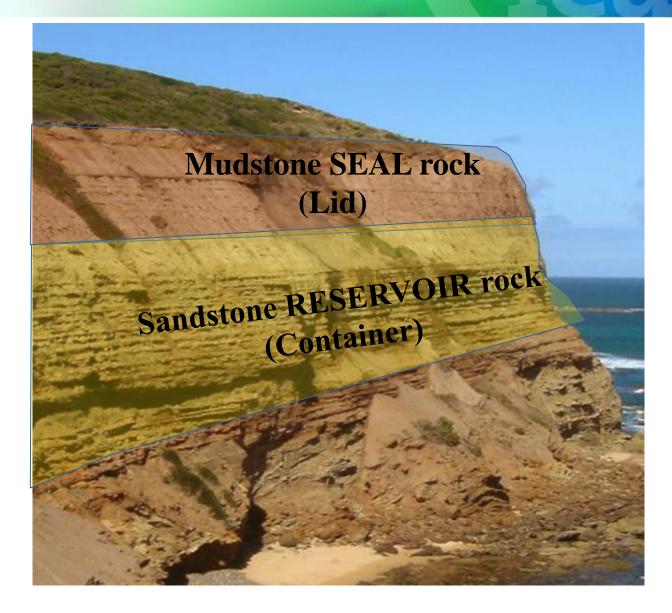
#### How can you store anything in rock?

The geological characteristics of the subsurface can be seen exposed in coastal outcrops

> Adapted from CO2CRC



Gamma Ray Curve



Adapted from CO2CRC

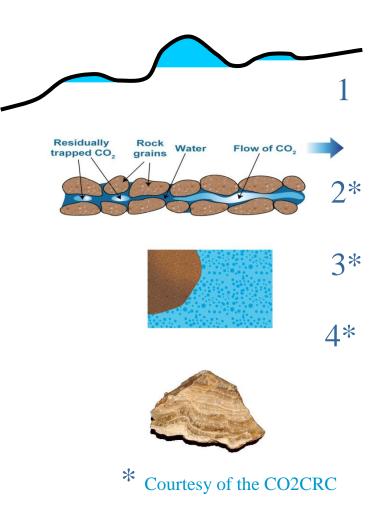


### **CO<sub>2</sub> trapping mechanisms in porous rocks**

When  $CO_2$  is injected into the subsurface, it will rise under buoyancy until it becomes immobilised by a combination of factors:

- 1. Structural and stratigraphic trapping
- 2. Residual trapping
- 3. Solubility trapping
- 4. Mineral trapping

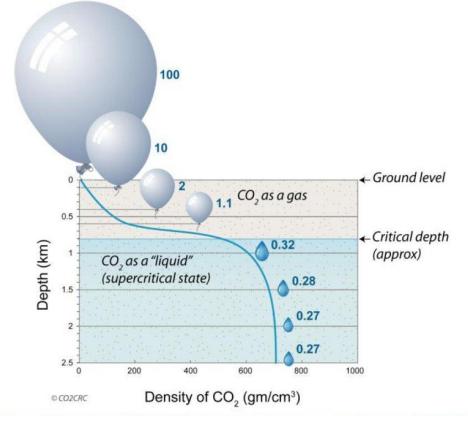
Unless residual storage occurs, the buoyant free phase  $CO_2$  will ultimately rise to accumulate under the top seal of the reservoir.





#### Why Supercritical CO<sub>2</sub>

- At Pressures higher than 7.39 MPa and Temperatures higher than 31.1°C, CO<sub>2</sub> becomes a supercritcal fluid: gas like but with 400x the density.
- Generally these conditions are found below about <u>800m</u> in the subsurface

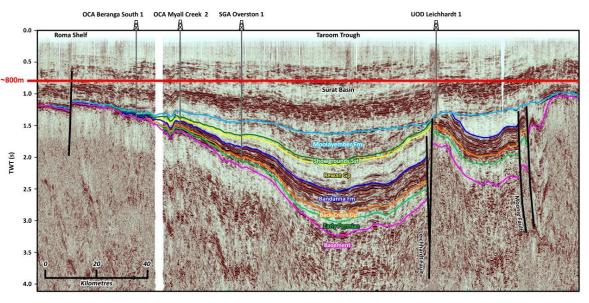


Source CO2CRC



#### Sedimentary basins and geological storage

- Saline aquifers suitable for storage occur almost exclusively in sedimentary basins
- These are depressions in the crust of the earth in which sediments have accumulated over millions of years and which have not experienced significant uplift and folding
- They may be tens of kilometres thick and occur both on the continents
- and under shallow seas
- All oil and gas accumulations occur in sedimentary basins.



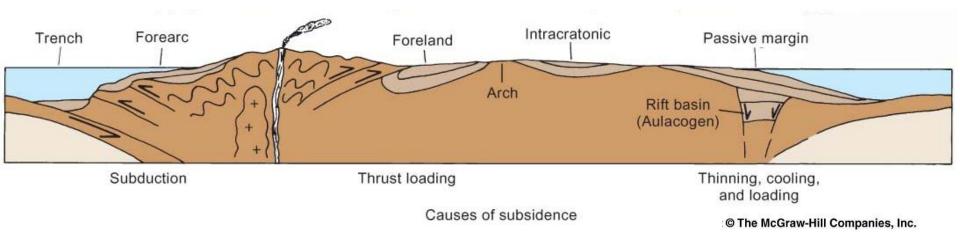
GSQ/GGSS Queensland Carbon Dioxide Geological Storage Atlas 2009.

 Oil and gas accumulations in a basin help demonstrate that the necessary trapping conditions exist there



# All Basins Are <u>Not</u> Equal

- Sedimentary basins are the regions that offer the opportunity for geological storage of CO<sub>2</sub>.
- But all sedimentary basins do not have the same potential for storage
- We need to consider the tectonic settings and reservoir characteristics of each basin un assessing storage

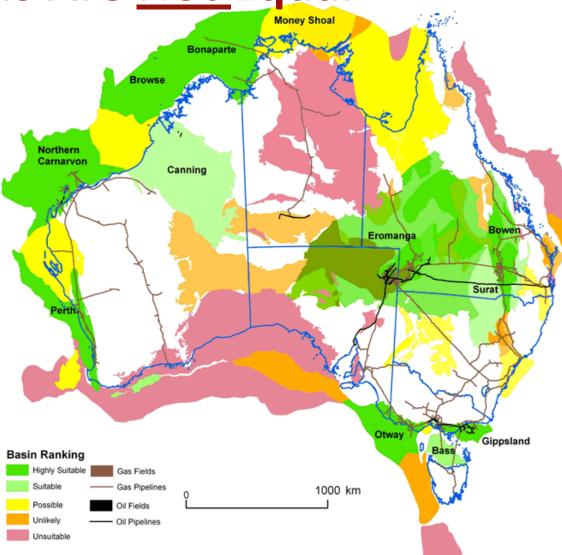




An example of the ranking of basins for carbon dioxide storage from a recent Australian Government Study conducted by Geoscience Australia

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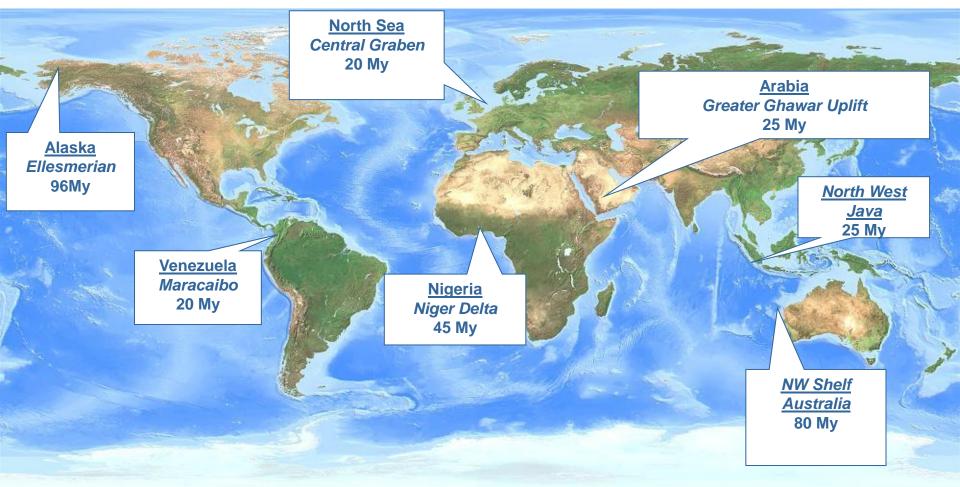


## How long will it stay there?

- Naturally occurring fluids have been trapped underground for many millions of years
- Oil, natural gas and CO<sub>2</sub>
- This can be shown by the study of petroleum systems.



#### **Time Of Petroleum Charge Into Traps**



UECD/IEA ZUIZ



#### Is This New Or Unproven Technology?

- The critical components of the CCS process are currently in use within the Oil & Gas Industry.
- Capture: Natural gas processing, ammonia plants other industrial processes.
- Transport: 5650 km of CO<sub>2</sub> pipeline in the USA.
- Injection: EOR 70 projects in West Texas. Acid gas disposal
- Storage: Subsurface storage of natural gas for 100yr storage of CO<sub>2</sub> since mid 1990s
- CO<sub>2</sub> storage in the North Sea since 1996



Source IPCC



## Thank You For Your Attention



#### **Pores not Caverns**

